

Application of TiO₂ and its derivatives for alternative energetic sources

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Abstract: Nowadays, the continuous rise of the population in the world emerges diversification of the energetic sources, for reliable energy supply. Furthermore, the sustainable development of the modern communities relies to environmentally friendly energy production equipment. In this aspect the present work has attention to overview some applications of TiO₂ as effective material for alternative energetic sources..

Key words: Titanium dioxide, Energy source, Photovoltaic cells, Water-splitting electrolyzers, SOFC,

INTRODUCTION

In nature TiO₂ exists in three basic polymorphic modifications: rutile, anatase and brookite, as is well described in ref. [1]. Nevertheless, various solid TiO₂- containing materials can be produced in artificial conditions, (i.e. industrial plants or laboratories). Furthermore, the respective TiO₂-based solids find large variety of applications for elaboration and production of different devices and equipment.

The purpose of the present brief review is to describe the main fields of TiO₂ application, remarking the potential abilities in the energy generation sources.

FIELDS OF APPLICATION

Titanium dioxide is well known as a photocatalyst and it is widely used for photocatalysed decomposition of various organic substances [2 - 5]. Recently, it has found various fields of application - for sensor elements, [6 - 10], water purification [11 - 12] and photodisinfection [11,12] systems, photoelectrochromic devices [13], electronic components [14, 15] and even for corrosion protection [16]. Furthermore, the beneficial properties of TiO₂ could be enhanced by its modification [17, 19]. Recently, this material has found effective for even medical applications, as medical implants [20 – 24] or medicine carriers and for gradual release of active substances [25 – 27], too. Due to its photo-activity, this oxide possesses a great potential for application for alternative energy sources. It could be used as layer in Dye Sensitized Solar Cells (DSSC) [28 - 32], Solid Oxide Fuel cells (SOFC) [33 - 35], or light induced water splitting devices [36 - 38]. The fields of application of TiO₂ containing materials are summarized in Fig. 1.

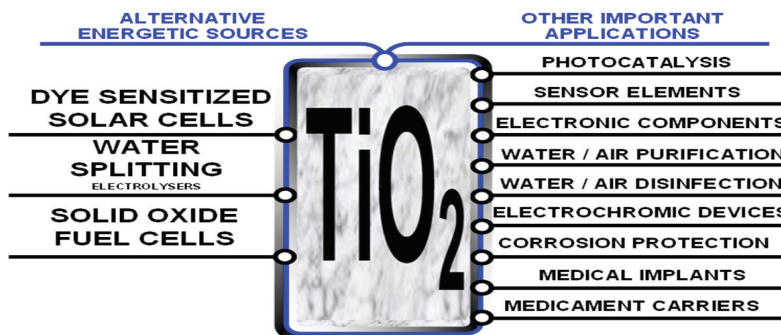


Figure 1. Fields of application of titanium dioxide solids

Undoubtedly, the characteristics of the TiO₂ containing devices depends on its structure, as well as on its form of application. This oxide could be obtained from large variety of solids with different structures and forms, according to the method and conditions applied for its synthesis. Examples of transparent, mesoporous anatase TiO₂

monoliths [39], highly dispersive powders [40 - 43], nanomaterials of TiO_2 with controlled porosity [44], dense [45 - 48] or porous [49 - 52] thin film coatings and others are found and a selection is presented here. The basic natural and artificial forms are illustrated in Fig. 2. Nanosize TiO_2 is not shown, since the field is too broad, innovative one and presents a separate object of overview and discussion.

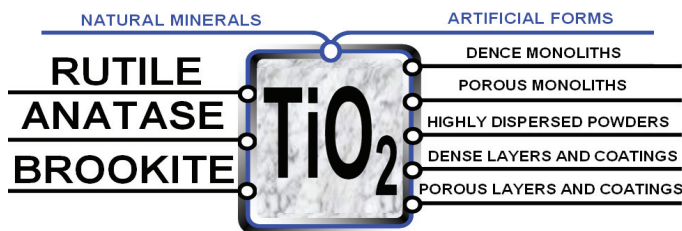


Figure 2. Natural and artificial titanium dioxide solids

CONCLUSIONS

The literature analysis for the present brief review enables to achieve the following conclusions:

1. Besides its use for alternative energy sources, TiO_2 has really extended fields of application, e.g. in the chemical industry, automation, electronics, environmental technologies, and medicine.
2. It could be produced in different forms, according to the methods and conditions, applied for its synthesis.
3. The authors of almost all of the articles, cited in the present brief review, have employed the sol-gel method, in order to obtain TiO_2 – based materials in various forms and with desirable properties. This fact reveals the significant versatility of the sol-gel approach of synthesis.
4. The spray pyrolysis techniques are also appropriated for TiO_2 powder production or film deposition, regardless the fact that the high temperatures required may cause polymorphic transitions during synthesis.

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